

DEPARTMENT OF PHARMACOGNOSY COLLEGE OF PHARMACY

FIRST SEMESTER (PHARM.D) EXAMINATION 2021/2022 SESSION

COURSE: PCG 301 Practical

Level: 300

Time Allowed: 2 hours (9:30-11:30AM)

INSTRUCTIONS:

1. Answer ALL Questions

2. Answer all the questions in your answer booklet

3. Write your Matriculation number in the space provided in your answer booklet

QUESTION 1

Observe Sample A and B and describe the shape, apex and margin

QUESTION 2

Describe the Specimen under the microscope with the aid of a well labelled diagram

QUESTION 3

- 1. Weigh 20g of Sample C given and spread on a clean white paper
- Inspect with unaided eye and pick out any foreign matter (other than leaflets) from the sample
- 3. Weigh the unwanted material

DEPARTMENT OF PHARMACEUTICAL AND MEDICINAL CHEMISTRY COLLEGE OF PHARMACY

AFE BABALOLA UNIVERSITY

2022/2023 Session; First Semester

PCH 301: Organic Pharmaceutical Chemistry I (Practical Examinations) TIME ALLOWED: 3 Hours

24th JANUARY, 2023

GROUP: B

Matric Number:.

ATTEMPT ALL QUESTIONS

[Total: 40 marks]

- Sample E is a mixture containing two chemical substances. Carry out the assay below to recover the desired substance.
 - vi. Weigh 1.8 2.0 g of sample E into a 250 mL conical flask and label as such. Add a pinch of boiling chip. Add 50 mL of hot water to the mixture and swirl.
 - ii. Place the flask on a hot place and boil for 5 minutes.

iii. Remove the flask from the hot plate and allow it to cool.

- iv. Place the flask in an ice bath for 5-10 min to ensure complete crystallization.
- v. When the flust, to cold and crystallination is complete, filter the crystals by suction.
- vi. Wash the crystals with 5 mL of ice-cold water, and repeat once.
- vii. Dry the crystals in the even at 50 °C and calculate the percentage yield of your dried product.
- Carry out the thin layer chromatography (TLC) of the product using 10 mL of sample F as mobile phase and calculate the Rr.
- ix. Submit your product labeled with your matriculation number at the end of the experiment.
- State four precautions taken during the experiment. b.
- What is the role of boiling chips in the experiment? -> \

2. [Total: 20 marks]

- A batch of paracetamol tablets produced by Drugtield Pharmaceuticals Limited was suspected to be fake. As a Quality Control Analyst, explain how you will ascertain the authenticity of the raw powdered drug. (Melting point of pure paracetamol: 169 - 171°C).
- What will be your recommendation to avoid any future occurrence? b.
- Write three (3) factors that determine the choice of solvent in crystallization technique. C.

18. The addition of a different substance to a drug in lieu of the actual certified component is called DU 60 to tutto co

19. A high ash value indicates. Corttomus cotton.....

20. High moisture content in crude drug prevents deterioration by aiding microbial growth, True or False

Section B imode, Ratioful, Sici, Temperature, Humiday, Altaude, Logisto of Haday 1a. Mention Seven (7) factors affecting Cultivation

b. Briefly discuss any two of the factors stated above.

- 2. What is the Importance of studying Enzymes in Pharmacognosy?
- Describe Papain and state its uses

4. List Five (5) Physical Parameters used in the Evaluation of Crude Drugs

- What asb.

- Extractive ralue

-Moisture Content

biological Paramolers.

Contamination

TOTAL VIable COUNTY

- Pharmacological paramoters

Parameters

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we corder leads to the sportage due to determination and

A.

B.

C.

D.

E.

F.

G.

H.

I. J.

2. Study the table below and answer the questions that follow.

Disinfectant	Dilution	Growth of test organism in subculture after exposure for:					
		2.5 mins	5 mins	7.5 mins	10 mins		
Test	1:400	NG	NG	NG	NG		
Disinfectant	1:500	G	NG	NG	NG		
	1:600	G	G	NG	NG		
	1:700	G	G	G	NG		
	1:800	G	G	G	G		
		The state of the s	Contract of the second		100000000000000000000000000000000000000		
	1: 95	G	NG	NG	NG		
	1:100	G	G	NG	NG		
Phenol	1:105	G	G	G	NG		
	1:110	G	G	G	NG		
	1:115	G	G ·	G	G ·		

- i. Calculate the phenol coefficient of the test disinfectant after 7.5 minutes.
- ii. What does your answer indicate?

3. Study the table below and answer the questions that follows.

In - Use 1057.

Concentration	Inoculum		Challenge 1	Result	
of test disinfectant	count	1	2	3	
1.0	2 x 10 ⁹	+++++	++++	++++	Contraction of the Contraction o
1.5	2×10^9	+	++++	++++	14
2.0	2×10^9	+	++	+++++	
2.5	2×10^9			+	

- i. Write the result for each of the concentration using either Pass/fail.
- ii. What is the name of the test represented with the table above?

DEPARTMENT OF PHAMACEUTICS AND PHARMACEUTICAL TECHNOLOGY COLLEGE OF PHARMACY

300 LEVEL PHARM D. FIRST SEMESTER EXAMINATION 2021/2022 SESSION

COURSE: PCT 301 (PHYSICAL PHARMACEUTICS I)

DATE: Monday 7th February, 2022.

TIME: 9.00am - 12noon

12-125

SECTION A

INSTRUCTIONS: Answer all questions.

- 1. a. List all possible interfacial boundaries that will exist between two immiscible liquid mixtures.
 - b. What is the difference between adhesive and cohesive force?
 - c. Explain why surface tensions is greater than interfacial tensions.
- a. Mention the bases of the Ostwald classification of Colloids and identify the resultant classes.
- b. List five kinetic properties of Colloids.
- 2 c. Explain the concept of solubilisation.

SECTION B

INSTRUCTIONS: Answer any four (4) questions only.

- 1. a. Distinguish Colloidal dispersions from other dispersions in the disperse systems.
 - b. Enumerate the methods you would employ in preparation of lyophobic sols.
 - . c. List six colloidal drug delivery systems you know.
- 2. a. Itemize with examples the pharmaceutical applications of Colloids.
 - 2 b. Identify three possible instabilities of colloidal dispersions and suggest two possible ways to counteract these instabilities.
- 3. Discuss the application of surface active agents in pharmaceutics.
 - 4. Explain the implications of stronger adhesive forces of attraction compared with cohesive forces.
- 5. Tabulate the differences between physical adsorption and chemisorption.





DEPARTMENT OF PHARMACEUTICAL AND MEDICINAL CHEMISTRY COLLEGE OF PHARMACY, AFE BABALOLA UNIVERSITY

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2021/2022 Session First Semester Examination PCH 301

TIME ALLOWED: 3 Hours

31st JANUARY, 2022

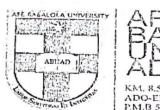
ANSWER QUESTION 1 AND ANY THREE (3) OTHER QUESTIONS. NOTE: Question 1 is compulsory.

1. (a) Predict the products in the following reactions and suggest plausible mechanism for each case

(i)
$$+$$
 CH_2 —CI $AICl_3$? (ii) R R $NaOH$?

- (b) Phenobarbital, a weakly acidic drug with pKa 7.4, is a barbiturate used as hypnotic and an anticonvulsant. What percentage of an orally administered dose of phenobarbital will exist in neutral (i.e. unionized) form in the:
 - (i) plasma (pH=7.4)
 - (ii) stomach (pH=1.4)
- 2. Use the chemical structures of the following five compounds to explain why each compound has the indicated pKa value. AND unambiguously identify each compound as acidic, basic, neutral or amphoteric:

- 3. (a) Starting with an appropriate alkyl halide and using any other needed reagent, outline the synthesis of 1-propanol. Explain the reaction mechanism.
 - (b) Explain the terms S_N1 and S_N2 in connection with the mechanism of the hydrolysis of primary, secondary and tertiary alkyl halides.



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COLLEGE OF PHARMACY

DEPARTMENT OF PHARMACEUTICAL MICROBIOLOGY

COURSE TITLE: PRINCIPLES OF STERILIZATION AND ASEPSIS

COURSE CODE: PMB 301

FIRST SEMESTER EXAMINATION FOR 2021/2022 SESSION

Date: 28/01/2022

Time: 30 Mins

SECTION A: MULTIPLE CHOICE QUESTIONS

Instruction: Answer all questions in this section by circling the correct answer for each of the question, write True or False where it is necessary and fill in the correct answer for the gap.

- 1. The process which removes visible contamination and a prerequisite for effective disinfection or sterilization is
 - a. Asepsis
 - b. Cleaning
 - c. Antiseptic
 - d. Sterility
- 2. The process which reduces the number of viable microorganisms to an acceptable level but may not inactive some viruses and bacterial spores is termed
 - a. Sterilization
 - b. Disinfectant
 - c. Decontamination
 - d. Sanitizing
- 3. The process that reduces microbial population on object to a safe level
 - a. Sterilization
 - b. Disinfectant
 - c. Decontamination
 - d. Sanitizing
- 4. The process that removes pathogenic microorganisms from an object to make it safe to handle
 - a. Sterilization
 - b. Disinfectant
 - c. Decontamination
 - d. Sanitizing
- 5. Which of the following statement is true of the objective of Presterilization cleaning?

. ;

- a. Removal of the organic matters, blood and saliva which provide protective barrier for microorganisms and prevents its destruction
- b. The destruction or removal of all microorganisms from article, surface or medium, including spores
- c. None of the above
- d. All of the above
- Classification of the physical method of sterilization//disinfectant include the following except
 - a. Irradiation
 - b. Filtration
 - c. Dyes
 - d. Gas
- 7. Autoclaves, or steam sterilizers essentially consist of the following except:
 - A non-cylindrical or octahedral chamber, with capacities ranging from 400 to 800 millilers.
 - b. Water heating system or steam generating system
 - c. Steam outlet and inlet valves
 - d. Thermometer or temperature gauge
- 8. Which of the following is not peculiar to pharmaceutical microbiology?
 - a. Sterilization
 - b. Asepsis
 - c. Antimicrobials
 - d. Hygiene

Answer True or False

- 9. Mechanical washing uses the principle of conversion of electrical energy into vibratory sound waves which pass through a soap solution containing the instrument
- 10. Ultrasound cleaning uses the principle of high-pressure jets of water with or without a detergent which removes debris from instrument.
- 11. In pasteurization, the temperature employed is either 63°C for 30mins or 72°C for 15-20 seconds followed by cooling quickly to 13°C.
- 12. In Chemical Indicator of sterilization control of the moist heat, an alloy is designed to melt only after being subjected to relevant holding time.
- 13. If the bioburden is high, long exposure times or higher concentrations of antimicrobial may not be required.
- 14. Many materials such as fabrics, rubber, plastics can adsorb certain disinfectants, e.g. quaternary ammonium compounds.

1.	
Fill in the g	ар
15. In sterili	zation process the different types of filter use in Filtration include Canala
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16. <u>memb</u>	
0	
17 Asb	(2)05
18. Sinter	
10	ca 916.50
19. The two t	ypes of non-ionizing Radiation used for sterilization are Interact
and	
20 1	
20. Ultravia	ol-c f
21 All drug n	reparations must be sterile.
	True ~
	False
	Not sure
	Don't know
	administered parenterally must be sterile.
	True V
b.	False
c.	Not sure
d.	Don't know
23. All drugs to	o be administered into internal organs must be supplied sterile.
a.	True /
Ъ.	False
c.	Not sure
d.	Don't know
24. Body fluids	s are excellent media for microbial growth.

- 24 a. True
 - b. False
 - c. Not sure
 - d. Don't know
- 25. Sterile preparations are..... of/from living microorganisms.
 - a. full
 - b. free
 - c. contained
 - d. made
- 26. Injections are packaged inside.....
 - a. bottle
 - b. container

c.	vial
d.	cylinder
	are administered with the aid of
	vials
	ampoule
	needle and syringe
	ligature
	ns administered into the eyes are
	injectable solutions
	irrigation fluids
	ophthalmic solutions
	dialysis solutions
	are used in poisoning of renal failure to remove accumulated waste or toxic
products	inicatable calutions
	injectable solutions
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	ophthalmic solutions dialysis solutions
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	suture
	ligatures /
	implants
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21 Strondonio	sed to stitch wounds in surgery are
	ligatures Sintered Sintered
D.	indicate a small
C.	surgical dressings
d.	solutions can cause discomfort when administered to the eye.
	True
	False
	Not sure
d.	Don't know
	as administration of solution can cause haemolysis.
	isotonic
b.	hypertonic
c.	hypotonic
A	mesotonic
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	hypotonic
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35	solutions can be used in selectoris.
	isotonic
√b.	hypertonic
C.	hypotonic
d	mesotonic dissociation in water?
ac What in the	mesotonic e dissociation factor of potassium chloride, having 60% dissociation in water?
36. What is the	UISSOCIA

a.	2.6
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37. Calculate the dissociation factor of magnesium sulphate, having 80% dissociation in water?

38. What is the dissociation factor of sodium chloride, having 60% dissociation in water?

Nacl

39. What is the dissociation factor of sodium carbonate, having 60% dissociation in water?

40. In adjusting isotonicity of solution, calculations may be based one of the these:

- a. Freezing Point Depression
- b. Boiling Point Lowering
- c. Evaporation
- d. Boiling Point Depression

SECTION B: THEORY OUESTIONS (20 marks each)

Instruction: Answer four questions in all, two questions from each part.

P	a	r	t	A
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What is Microbial death? Support your answer with graphical illustration. 1(a).

The intended application of an antimicrobial agent, whether for preservation, 1(b). antisepsis or disinfection, will influence its selection and also affect its performance, Discuss.

Mention 7 antimicrobial agents/disinfectants and give a detail explanation of any 3 you understood.

2(b). Briefly explain 3 mode of action of antimicrobial agent

Enumerate 5 factors affecting choice of antimicrobial agent-2(c).

10 . i elticals of citimise copial about to pe What is Microbiological challenge in estimated by o propriate measures to ensur 3(a).

Describe the detailed Microbiological challenge of Vegitative bacteria, support your that a store answer with appropriate examples. , d + = obtaine which is appropriate to the

to used for disynfection should be calable of Killing most estandarduse. it also Part B vegetative bacteria within a resemble compet

ne cessary to Catego List and explain any five classes of sterile preparations. chemicals according List and explain all the types of injections based on volume of injection. to the tr

The danger posed by injecting paratonic solutions depends on the route of

administration and volume of injection. Explain in detail, the consequences of injecting paratonic solution through the following routes:

a. Intravenous route.

(2) b. Intrathecal route

- 5 c. Intramuscular route
- 4 d. Intradermal route
- (3) e. Opthalmic route

- microbial thallinge - Course mental factors Define Sodium Chloride Equivalent. 3(a).

Find the quantity of sodium chloride to be used in compounding the following prescription. The values of sodium chloride equivalent of glycerin and scopolamine · broberius of q eperminal whom hydrobromide are 0.34 and 0.12 respectfully.

70.309.

Rx

Glycerin 1%

Scopolamine HBr 0.5%

Sodium chloride q.s

Purified water ad 60.0

Radiation of St Make isotonic solution. - Hullsice cid 3(c). Find the quantity of boric acid (in grams) to be used in compounding the following

chemicals may damage the Dr prescription:

Atropine sulphate 0.5%

Boric acid q.s o. 52

Purified water ad 30.0

Mary ISOHOMIC SOLUTION

Boric acid C.51

two Hoyming wises.

26) mode of action of a colone

- Introded application

26) - Proleins: Some Proleins are donatu

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DEPARTMENT OF PHARMACOLOGY & TOXICOLOGY COLLEGE OF PHARMACY

FIRST SEMESTER 2020/2021 SESSION PCO 301 THEORY

INSTRUCTIONS: Answer three (3) questions in all

TIME ALLOWED: I hr

SECTION A [COMPULSORY]

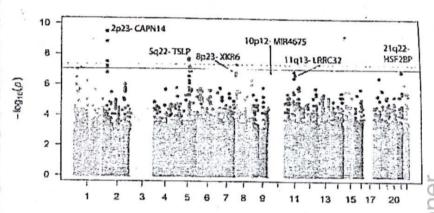
Eosinophilic esophagitis (EoE) is a chronic inflammatory disorder associated with allergic hypersensitivity to food. In the study below, Kottyan et al. interrogated >1.5 million genetic variants in EoE cases of European ancestry and control subjects. The plot below is the result of that study.

Q1:What is the name of this type of study (0.5 mk)? Q2:What is the name of this plot (0.5 mk)?

Q3:What is the name of the X-axis of this plot (Imk)?

Q4: Identify the most plausible drug target for EoE from the graph and identify the chromosome where the gene can be found (1 mk)

mepolizumab (monoclonal antibody against IL-5)



Q5: CAPN14 and its gene product CAPN14 is an important member of the calpain family; they are intracellular, calcium-dependent cysteine proteases and its emerging roll in patients with EoE. As a cysteine protease, name the three amino acids you would expect in the active site of this enzyme and by what name are these amino acid groupcalled (1 mks)? 1413 Fre

Q6:The compounds above are both CAPN14 inhibitors with experimentally demonstrable activities in EoE, which of these two compounds is a reversible inhibitory (I mk)?

Q7: Which is the irreversible inhibitor (Imk)

Q8: What is the general name given to head group of such irreversible inhibitors (1mk)?

Q9: Comment on the chemistry underlying the reaction between the head group and the active site amino acids (1 mk)

Q10: Justify the answers in Q7 and Q8 above (1 mk)

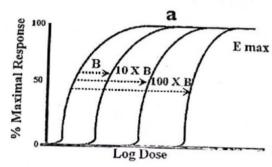
Q11: Since CAPN14 is a cytosolic calcium-activated cysteine protease, comment on (a) class of drug target CAPN14 is, (b) the most appropriate route of administration for all the drugs (i-iii) above that allows efficient tissue distribution and cellular uptake, based on their physicochemical properties (5 mk)

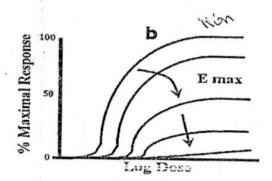
Q12: List out other types of drug target there is based on cellular location (3 mks)

Q13: Calpain activity assay protocol is based on the detection of cleavage of calpain substrate Ac-LLY-AFC. Ac-LLY-AFC emits blue light (\lambda max = 400 nm); upon cleavage of the substrate by calpain, free AFC emits a yellow-green fluorescence (\lambda max = 505 nm), which can be quantified using a fluorometer or a fluorescence plate reader. Draw a hypothetical graph that shows inhibition of Calpain-14 in the presence of compounds "i" and "ii" above in comparison to a vehicle control (3) mks) maliante

SECTION B [ANSWER ONLY ONE QUESTION FROM THIS SECTION]

1





- ai. Identify and name graphs a & b above (2 mk)
- ii. In a tabular form, give at least 3 difference between graphs a & b above (5 mk)
- iii. Give at least one example each of drugs that belong to graphs a & b above. (2 mk)
- b. List at least 5 factors affecting drug absorption (6 mk)

2 20 a

- ai. If the y-axis represents plasma concentration (mcg/ml) and x-axis represents time (h) on the graph, identify a pharmacokinetic parameter that can be gotten from the graph (3 mk).
- ii. Name the routes of administration represented by 1 & 2 in the graph (2 mk) 1x \$ 0001
- iii. List 4 factors affecting the identified pharmacokinetic parameter (4 mk)
- b. Briefly explain 3 parameters that can be gotten from the all or none dose response curve (6 mk)

SECTION C [ANSWER ONLY ONE QUESTION FROM THIS SECTION]

- Ta. Describe in details forensic toxicology (10 mk)
 - b. Explain in details factors affecting the effects of poisons (5 mk)
- 2a. Enumerate the classification of poisons (5 mk)
- b. Describe the factors affecting the delivery of poison (10 mk)

D-Chief Tharmocist



TI (ABUAD)

COLLEGE OF PHARMACY Department of Clinical Pharmacy & Public Health

First Semester Examination 2021/2022 Academic Session

Course: SAP301: Pharmacy Administration

Exam date: Thur 10/02/22 9.00am - 12.00noon

SECTION B (ESSAY – Attempt any FOUR questions)

- You have been contacted by a firm with limited funds who wishes to partner with you in establishing a 1. pharmacy enterprise around your locality in Nigeria.
 - a) Outline the sources of viable funding you will recommend and the limitations of each?
 - b) Highlight in tabular form, the factors you would consider in choosing between building a pharmacy from scratch or buying an already existing pharmacy [11 marks, ENO]
 - 2. (a) List the operational principles of good drug procurement
 - (b) List and explain the drug procurement methods that can be used to aid the procurement process
 - (c) Highlight 5 challenges that can hinder the procurement process? [11 marks, ENO]
- 3. Consultants are hired to share their expertise and knowledge to help businesses attain goals and solve problems.
 - (a) Outline six (6) qualities of a good consultant
 - (b) List six (6) areas where a pharmacist be a consultant .
 - (c) If you were appointed a consultant by a private company, outline all possible ways to determine the consulting rate [11 marks] [POE]
- (a) Define marketing communication.
 - (b) Outline the importance of (marketing communication •
 - (c) Outline the characteristics of excellent marketing skills [11 marks] [EOO]
- 5. (a) Define entrepreneurship development
 - (b) Outline the steps in conducting market survey.
 - (c) Outline 7 steps in selecting a business location [11 marks] [EOO]
- 6. Outline the stepwise procedure to start a new pharmacy practice [11 marks] [EOO]

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DEPARTMENT OF PHARMACOGNOSY COLLEGE OF PHARMACY

FIRST SEMESTER (PHARM.D) EXAMINATION 2021/2022 SESSION

COURSE: PCG 301

Level: 300

Date: 2nd February 2022

Time Allowed: 3 hours (9:00- 12:00AM)

INSTRUCTIONS:

- Answer ALL questions in Section A and any three (3) in Section B
- Answer all the questions in your answer booklet
- Write your Matriculation number in the space provided in your answer bookiet

SECTION A

- 1. Cultivation ensures quality and purity of medicinal plants. Truc/False
- 2. Show how altitude and rainfall affect cultivation
- 3. Name the two methods of propagation of medicinal plants.
- 4. In sexual propagation, seedling is comparatively cheaper and easy to raise. I'rue/False
- 5. Resins are Amarphous or Some solid
- 6. Tolu balsam is a balsam obtained from Tolucifera balsamum family. പ്രവസാട
- 7. The major component of the oil of citrus fruit peels is. A. Limprone
- 8. The instrument used in extraction of Volatile oils is called Colonger apparation.
- 9. Enzymes obtained from plants are called Photo encymes [Person Remain]
- 10. Enzymes derived from animals are known as Rancelotic engines
- 11. Drug Evaluation is the determination of identity, purity and quality. True/False
- for cell features searched under 12. Examples are Trichomes and fragmorded epidermal couls.
- 13. What is the full meaning of VIN and VTN?
- 14. Substituting real ginger powder with exhausted ginger is a type of unintentional adulteration. Truc/False
- 15 Pure bocce- is the major substitute for original honey in adulteration
 - 16. The toxic part of Abrus precatorius is . seed 500d
 - 17. All parts of Nicotiana tabacum are poisonous.

True/False

- 18. The inherent toxicants found in plants are secondary metabolites. True/False
- 19. Plants produce toxicants as part of their self defence system against pathogens and stresses. Truc/False
- 20. Toxins of Abrus precatorius are released when seeds are chewed and swallowed. True/False
- 21. Cannabis refers to a group of The QL... plants with Policies properties.

VIII - Vincostine & Vinblostine canned with CamScanner

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cannabis tuderalis

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Maida anni	. 22			he "high" that m					
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			/False	L 10 William Primi		1			
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				stem (b) dry we		(c) damp weathe	r; wood (d) we	et weather;	
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				es (d) Open fires					
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1		1.		ferent methods		on •	COLORI -		
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afacon		c.	Describe th	e Chemical test	ts for Volati	le Oils		Brangleat	
200		d.	Mention for	ur methods of a	dulteration	1		CHEMICAL	e.
mous cal					. 7 2.	· de			
moderal	r4.					1991	. 1		
	:	a.	Outline me	thods used for p	propagation	by vegetative n	neans t		
		L		ore purposes of					
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Just									
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Oc.									

Admirture

- 10). Certain plants e.g. Althaea Officinalis that produce mucilage as a water-retaining material contain less muchage when grown on soil with Moisture content
- 11). Soils containing much humus and little lime are inclined to become. Acidic..., while those with abundant lime are Alkerting....

- 14). Drying process varies in duration from few hours to Many weeks.. and in case of open-air.
- (15). Leaves and other delicate structures when over become
- 17). Fibres may be defined as any their like raw material directly obtainable from an Animal vegetable or Mineral source.

 - 19). Chemically, all vegetable fibres consist mainly of ... Cettucose., although they also contain varying amounts of substances as temicanuose, lighth. Rections... and waxes that must be removed or reduced by processing.
 - 20). Dressings are compulsorily needed for Proper management and subsequent Healing of wounds caused by injuries, burns, microbial infections and surgical operations.



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COLLEGE OF PHARMACY

19 mHS 116 Department of Clinical Pharmacy & Public Health

First Semester Examination 2021/2022 Academic Session

Course: PCL301: Biopharmaceutics and pharmacokinetics Exam date: Mon 14/02/22 2.00 - 5.00pm

SECTION B (ESSAY - Attempt any FOUR questions)

a) List 5 physiological and 5 physicochemical factors that affects drug absorption [5 marks]

b) Explain (with relevant examples) how each of any 3 physicochemical factors listed above affects drug absorption [6 marks] {ENO} [1 hong] [10 hong]

2. Explain the following:

- a. Continuous increase in the plasma concentration of amoxycillin, in the first 25 mins, following oral administration and subsequent continuous decline in the concentration [3 marks]
- b. Toxicity of a drugs with a half-life of 20 hr administered four times a day [3 marks]

c. Toxicity of a drug in a patient whose creatinine clearance decline by 75% [2 marks]

- d. Why the plasma concentration profile of amoxycillin 500mg and 2000mg remained relatively similar following oral administration of the drug [3 marks]. {POE}
- 3. (a) Differentiate between active transport and passive diffusion [5 marks] (b) Outline the characteristics of facilitated diffusion [6 marks] {POE}

Outline how you will determine the following parameters? 4.

a. Absolute and Relative bioavailability of solid dosage formulation for oral administration [5 marks].

b. Biocquivalence of a new solid dosage formulation for oral administration with a similar innovator drug formulation [6 marks]. {POE}

(a) Define bioavailability of a drug. [2 marks]. 5.

(b) Give three reasons why bioequivalence studies are required for a new solid dosage form before it

is registered by the regulatory body [3 marks].

(c) The bioavailability of a drug whose volume of distribution (V_d) is 40 litres and half-life is 14 hr is 70%. If 300mg is administered orally at a dosing interval of 18 hr, calculate the steady-state plasma concentration (C_{ss}) for the drug $\{C_{ss} = F \cdot (D/t)/(k_{el} * V_d)\}$ where F is the fraction of the drug absorbed, D is the dose administered, t is the dosing interval and V_d is the apparent volume of distribution [6 marks]. {POE}

Consider the following values obtained after the oral administration of 300mg of phenytoin tablet. 6.

Time (hr)	1	2	5	10	15	20	20	40	
$C_p (\text{mg/L})$	0.6	2.0	1	110	15	20	30	40	50
7 (-8-)	0.6	2.0	. 4	4.10	3.6	3.3	2.1	1.2	0.70

(a) Calculate the AUC_{0-15hr} and AUC_{0-50 hr} [6 marks].

(b) Given that the elimination rate constant (ke) for the phenytoin is -0.048/hr and half-life (t½) is 0.693/ke, calculate the t1/2 of the drug [5 marks]. {POE}

DEPARTMENT OF PHARMACEUTICS AND PHARMACEUTICAL TECHNOLOGY

PCT 316: Dispensing Practical IV 2022/2023 Session (Second Semester)

TEST

Wednesday	May	17,	2022
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Time Allowed: 1 Hr 30 Mins

REG. NO.:	Te day
	2

Answer ALL Questions in Sections A and B GROUP A

SECTION A

INSTRUCTIONS1: The preparation of the products is required in Section A. The attention of the Examiner must be drawn to any unusual doses and irregular labels. All quantities used and dispensed and all calculations must be clearly shown. The method of preparation should only be given where required.

1. R_X

Prepare 20 g of Methyl Salicylate Ointment PCx 1979. Label: Apply to the inflamed skin area twice daily.

*State the method of preparation

2. R_X

Triethanolomine	0.25 g
Gleic acid	1.0 g
Benzyl Benzoate	12.5 ml
Water	12.5 ml
Water to	100%
	nladi Usman and Label Appropriately.

SECTION B

INSTRUCTIONS: Preparation of the products is not required in this section. However, you are required to show your calculations ONLY.

3 a) R_X

> Fluconazole tabs 100mg

Disp. ____ tabs

Sig: tab ii stat, then 3mg/kg b.i.d. × 7 d thereafter

Calculate the number of tablets to dispense to a patient weighing 147 lb.

b) Rx

> Age: 4 years George Ighodalo

Weight: 52.8 pounds

Erythromycin Ethyl Succinate (EES) 200mg/5ml

Disp. 300ml

Sig: ____ ml qid until gone

i) If the dose of EES is 50mg/kg/day, how many milliliters would provide each

ii) How many days would the prescription last the patient?

SECTION B: Answer ANY TWO (2) Questions (20 MARKS EACH)

- You are given a conical flask containing 2000 mLs of water collected from ABUAD borehole, prepare and send 90 ml of bacteriostatic water for injection.
- You are provided with phenyl mercuric nitrate 0.02% as bactericide, a glass eye drop bottle
 and a sintered glass filter No 4. Send 10 ml of 0.25% w/v zinc sulphate eye drop B.P.C.
- 3. You are provided with 50g of anhydrous dextrose, 9g of sodium chloride and 1000 mLs of freshly distilled water, prepare and send 250 mLs of intravenous dextrose saline infusion 0.9%w/v

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COLLEGE OF PHARMACY

24th November, 2022

DEPARTMENT OF PHARMACOGNOSY AND NATURAL PRODUCTS

MID-SEMESTER TEST 2022/2023 SESSION

COURSE CODE: PCG 301

TIME ALLOWED: 60MINS

COURSE TITLE: PHARMACOGNOSY I

Instruction: Answer ALL Questions in Sections A and B.

Section A

- Cultivation ensures quality and purity of medicinal plants...... True or False Cultivation ensures irregular supply of crude drugs......True or False 4. Name the two methods of propagation of medicinal plants 500d5 (~ 0 garatido. 5. Gibberellic acid promotes the germination of dormant seeds....... True or False 6. What enzymes are indicated in the treatment of pancreatitis.....? 7. As a Pharmacist, what enzyme medication would you recommend for dyspepsia? 8. What is an autocatalytic enzyme? Give an example 9. Give two reasons why we rely on microorganisms for the production of enzymes in modern Pharmacognosy Pancreatic amylase is a β-amylase, True or False 11. Resins are AND PRODUS solid or semi solid substances that are Worthor insoluble but mostly soluble in allowed (a) amorphous, alcohol, water (b) non amorphous, alcohol, water (c) non amorphous, water, alcohol (d) amorphous, water, alcohol 12. are found to be hard, translucent or transparent and fusible. (a) Chemically, resins (b) Physically, oleoresins (c) Physically, resins (d) Chemically, oleoresins 13. Oleoresins are semi-solid extracts composed of resin and obtained by evaporation of the solvents used for their production. (a) Olive oil (b) Corn oil (c) Vegetable oil (d) essential or fatty oil
- 14. are prepared from e.g. ginger, basil, nut meg, thyme, balsam, Indian bay leaves etc. (a) Oleoresins, spices (b) Resins, exudates (c) Oleoresins, exudates (d) Resins, spices
- causes the body to make substances that fight pain and swelling.

(a) Bromelain (b) Malt extract (c) Papain (d) Oleoresin

CLEVENGER'S APPARATUS

16. NOIGHAR. D. Is used to standardize drugs which the efficiency is due to odourous principles

17. The apparatus used for volatile oil content determination is Clonongar. D. Apparatus microargarusm

2ER3 APPARATUS

200 Proumo Establication of P



BARALOLA ADO-ERITI (ABUAD)

KM, 8.5. AFE BABALOLA WAY, ADO-EKITI, EKITI STATE, NIGERIA. PM B 5454 ADO-EKITI.

COLLEGE OF PHARMACY

PHARMACEUTICAL MICROBIOLOGY DEPARTMENT

COURSE TITLE: PRINCIPLE OF STERILIZATION AND ASEPSIS

COURSE CODE: PMB 201

21st January, 2022

Continuous Assessment

ANSWER ALL

Time: 40 minutes

 To achieve sterilization of Instrument, Theatre and Pharmaceutical Laboratory, three definite stages are very essential to be completed. Discuss



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P.M.B. 5454, ADO-EKITI

COLLEGE OF PHARMACY DEPARTMENT OF PHARMACEUTICAL MICROBIOLOGY AND BIOTECHNOLOGY

300L FIRST SEMESTER (2021/2022) TEST

Answer All 30 Minutes

- 1. Define Isotonic Solution
- 2. Explain the consequences of intravenous administration of paratonic solutions.
- 3. Write short note on:
- i. irrigation fluids ii. implants iii. absorbable haemostatics Scanned with CamScanner



COLLEGE OF PHARMACY DEPARTMENT OF PHARMACEUTICAL MICROBIOLOGY AND BIOTECHNOLOGY

300L FIRST SEMESTER (2021/2022) TEST Answer All 30 Minutes

- 1. Define Isotonic Solution
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DEPARTMENT OF PHARMACEUTICAL AND MEDICINAL CHEMISTRY COLLEGE OF PHARMACY

PCH 301 In-course Assessment Test Instructions: Answer ALL questions

Time: 1 hour

- 1. Starting with an appropriate alkyl halide and using any other needed reagent, outline synthesis of 1-propanol. Explain the reaction mechanism. -6-6-6-04
- 2. Explain the following:
 - (a) Free radicals
 - (b) Electrophilic and nucleophilic reagents
 - (c) Homolytic and heterolytic fission of covalent bond.
- 3. Explain the mechanism of substitution reaction of chlorine on methane.

CH3CH2CH3 + OH CH3CH2CH3 + OH